

TitraVer Titration Method**Method 8214****10–1000 mg/L as Fe****Digital Titrator****Scope and application:** For water, wastewater and seawater.**Test preparation****Before starting**

The optional TitraStir Titration Stand can hold the Digital Titrator and stir the sample.

Review the Safety Data Sheets (MSDS/SDS) for the chemicals that are used. Use the recommended personal protective equipment.

Dispose of reacted solutions according to local, state and federal regulations. Refer to the Safety Data Sheets for disposal information for unused reagents. Refer to the environmental, health and safety staff for your facility and/or local regulatory agencies for further disposal information.

Items to collect

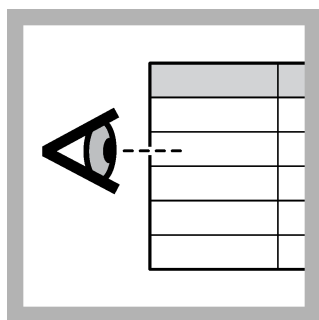
Description	Quantity
Citrate Buffer Powder Pillow	1
Sodium Periodate Powder Pillow	1
Sulfosalicylic Acid Powder Pillow	1
TitraVer Standard Solution Titration Cartridge (refer to Sample volumes and digit multipliers on page 3)	1
Digital Titrator	1
Delivery tube for Digital Titrator	1
Graduated cylinder (use a size that is applicable to the selected sample volume)	1
Erlenmeyer flask, 125 mL	1
Water, deionized	varies

Refer to [Consumables and replacement items](#) on page 4 for order information.

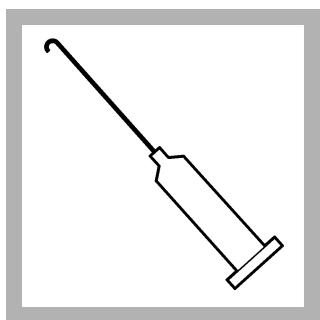
Sample collection

- Collect samples in clean glass or plastic bottles.

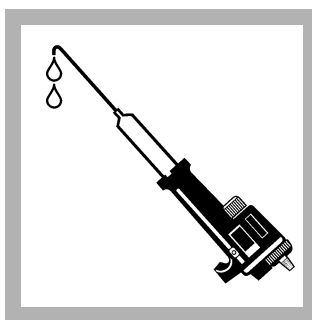
Test procedure



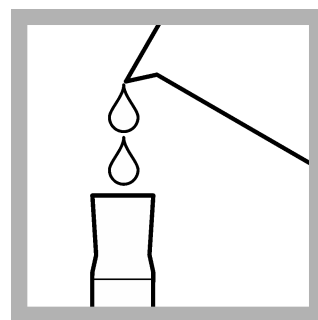
1. Select a sample volume and titration cartridge from [Table 1](#) on page 3.



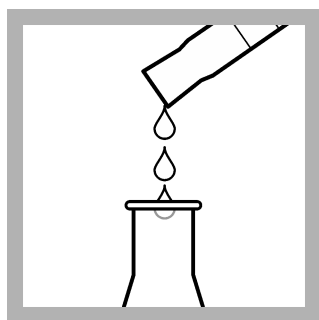
2. Insert a clean delivery tube into the digital titration cartridge. Attach the cartridge to the Digital Titrator.



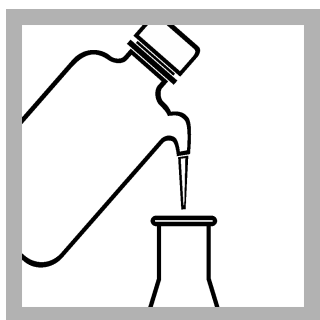
3. Hold the Digital Titrator with the cartridge tip up. Turn the delivery knob to eject air and a few drops of titrant. Reset the counter to zero and clean the tip.



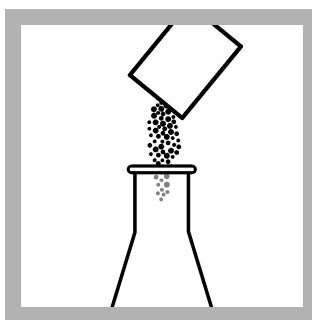
4. Use a graduated cylinder or a pipet¹ to measure the sample volume from [Table 1](#) on page 3.



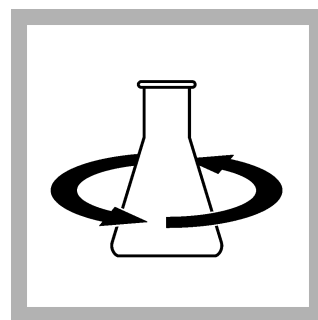
5. Pour the sample into a clean, 125-mL Erlenmeyer flask.



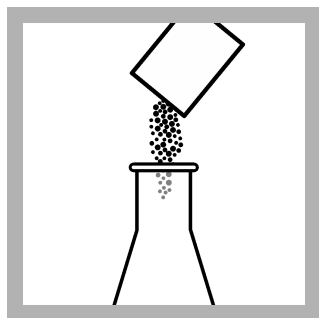
6. If the sample volume is less than 50 mL, dilute to approximately 50 mL with deionized water.



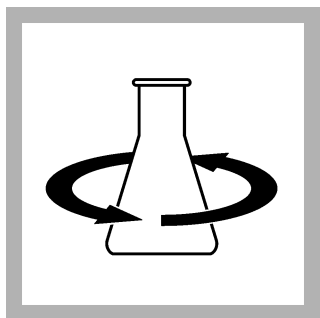
7. Add the contents of one Citrate Buffer Powder Pillow.



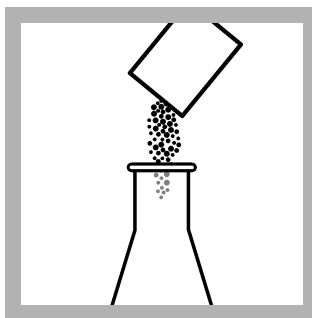
8. Swirl to mix.



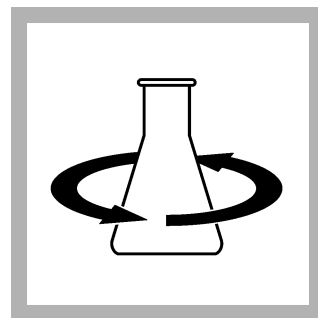
9. Add the contents of one Sodium Periodate Powder Pillow.



10. Swirl to mix. The color of the solution changes to yellow if iron is in the sample.

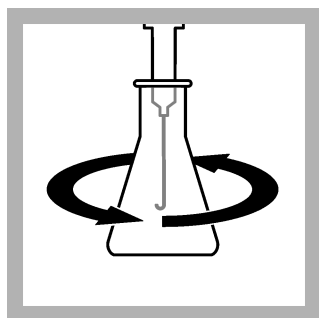


11. Add the contents of one Sulfosalicylic Acid Powder Pillow.

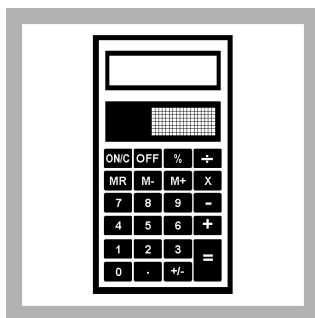


12. Swirl to mix. The color of the solution changes to red if iron is in the sample.

¹ Titration accuracy has a direct relation to the accuracy of the sample volume measurement. For smaller volumes, it is recommended to use a pipet to increase accuracy.



13. Put the end of the delivery tube fully into the solution. Swirl the flask. Turn the knob on the Digital Titrator to add titrant to the solution. Continue to swirl the flask. Add titrant until the color changes from red to yellow. Record the number of digits on the counter.



14. Use the multiplier in [Table 1](#) on page 3 to calculate the concentration.
 Digits used $\times$ digit multiplier
 = mg/L Fe.

Sample volumes and digit multipliers

Select a range in [Table 1](#), then read across the table row to find the applicable information for this test. Use the digit multiplier to calculate the concentration in the test procedure.

Example: A 50-mL sample was titrated with 0.0716 M TitraVer Standard Solution Titration Cartridge and the counter showed 250 digits at the endpoint. The concentration is $250 \text{ digits} \times 0.1 = 25 \text{ mg/L Fe}$.

Table 1 Sample volumes and digit multipliers

Range (mg/L as Fe)	Sample volume (mL)	Titration cartridge	Digit multiplier
10–40	50	0.0716	0.1
25–100	20	0.0716	0.25
100–400	50	0.716	1.0
250–1000	20	0.716	2.5

Accuracy check

Standard additions method (sample spike)

Use the standard additions method to validate the test procedure, reagents, apparatus, technique and to find if there is an interference in the sample.

Items to collect:

- Iron Standard Solution, 1000 mg/L as Fe
- Pipet, TenSette, 0.1–1.0 mL and pipet tips

1. Use the test procedure to measure the concentration of the sample.
2. Use a TenSette pipet to add 0.5 mL of the standard solution to the titrated sample.
3. Titrate the spiked sample to the endpoint. Record the number of digits on the counter.
4. Add one more 0.5-mL addition of the standard solution to the titrated sample.
5. Titrate the spiked sample to the endpoint. Record the number of digits on the counter.
6. Add one more 0.5-mL addition of the standard solution to the titrated sample.
7. Titrate the spiked sample to the endpoint. Record the number of digits on the counter.
8. Compare the actual result to the correct result. The correct result for this titration is 10 digits of 0.0716 M TitraVer Standard Solution Titration Cartridge (100 digits of the

0.0716 M titration cartridge) for each 0.5-mL addition of the standard solution. If much more or less titrant was used, there can be a problem with user technique, reagents, apparatus or an interference.

Summary of method

Ferrous iron Fe^{2+} is oxidized by sodium periodate to ferric ion Fe^{3+} . The ferric ion develops a red complex with sulfosalicylic acid. The red complex is removed by titration with EDTA. Citric acid is used to buffer the solution and to make the ferric ion stable in the solution.

Consumables and replacement items

Required reagents

Description	Quantity/Test	Unit	Item no.
Reagent set, 10–100 mg/L range (approximately 100 tests):	—	each	2449200
Citrate Buffer Powder Pillows	1 pillow	100/pkg	2081599
Sodium Periodate Powder Pillows	1 pillow	100/pkg	98499
Sulfosalicylic Acid Powder Pillows	1 pillow	100/pkg	2081669
TitraVer Standard Solution Titration Cartridge, 0.0716 M	varies	each	2081701
Reagent set, 100–1000 mg/L range (approximately 100 tests):	—	each	2449300
Citrate Buffer Powder Pillows	1 pillow	100/pkg	2081599
Sodium Periodate Powder Pillows	1 pillow	100/pkg	98499
Sulfosalicylic Acid Powder Pillows	1 pillow	100/pkg	2081669
TitraVer Standard Solution Titration Cartridge, 0.716 M	varies	each	2081801

Required apparatus

Description	Quantity/test	Unit	Item no.
Graduated cylinders—Select one or more for the sample volume:			
Cylinder, graduated, 25 mL	1	each	50840
Cylinder, graduated, 50 mL	1	each	50841
Digital Titrator	1	each	1690001
Delivery tube for Digital Titrator, J-hook tip	1	5/pkg	1720500
Pipet filler, safety bulb	1	each	1465100
Pipet, volumetric, Class A, 20 mL	1	each	1451520
Flask, Erlenmeyer, 125 mL	1	each	50543

Recommended standards

Description	Unit	Item no.
Iron Standard Solution, 1000 mg/L as Fe	100 mL	227142
Iron Standard Solution, 10 mg/L as Fe	500 mL	14049
Iron Standard Solution, 25 mg/L as Fe	10 mL/16	1425310
Iron Standard Solution, 50 mg/L as Fe	10 mL/16	1425410
Iron Standard Solution, 100 mg/L as Fe	100 mL	1417542

Optional apparatus

Description	Unit	Item no.
Ampule Breaker, 10-mL Voluette® Ampules	each	2196800
Pipet, TenSette®, 1.0–10.0 mL	each	1970010
Pipet tips for TenSette® Pipet, 1.0–10.0 mL	50/pkg	2199796
Stir bar, octagonal	each	2095352
TitraStir® Titration Stand, 115 VAC	each	1940000
TitraStir® Titration Stand, 230 VAC	each	1940010
Delivery tube for Digital Titrator, 90-degree bend for use with TitraStir Titration Stand	5/pkg	4157800



FOR TECHNICAL ASSISTANCE, PRICE INFORMATION AND ORDERING:

In the U.S.A. – Call toll-free 800-227-4224

Outside the U.S.A. – Contact the HACH office or distributor serving you.

On the Worldwide Web – www.hach.com; E-mail – techhelp@hach.com

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